# IDENTIFICATION OF URINARY METABOLITES OF RIFAMYCIN LM 427 IN MAN

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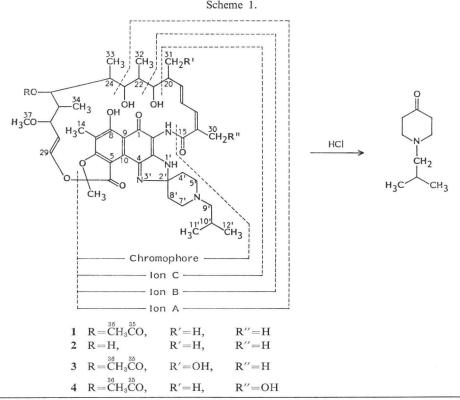
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LM 427 (1) is a new spiropiperidylrifamycin,<sup>1)</sup> with activity against atypical mycobacteria and many rifampicin resistant *Mycobacterium tuberculosis* strains.<sup>2)</sup> In the present paper the spectroscopic and chemical evidence leading to the identification of three metabolites (2, 3 and 4), present in human urine is described.

LM 427 was administered per os to six healthy

volunteers as a single dose (300 mg). The urines of each subject were collected during the first 24 hours after administration and pooled (total volume 8 liters). The compounds were isolated by extraction with a 9:1 mixture of EtOAc and 1-propanol, followed by preparative chromatography on silica gel plates (CHCl<sub>3</sub> -MeOH, 9:1). According to a semiquantitative evaluation of the TLC plates, the most abundant compound 1 is equivalent to the total of metabolites 2 and 3, which are almost in the same amount; on the contrary compound 4 is present in a smaller amount.\* Acid hydrolysis of 2, 3 and 4 (carried out at 40°C for 3 hours with 2 N HCl in acetone, followed by neutralization with NaHCO<sub>3</sub> and extraction with CHCl<sub>3</sub>) affords, as in the case of 1, the unchanged N-isobutyl-4-piperidone moiety, identified by GC-MS comparison with an authentic sample, thus ruling out any modification of this part of the molecule.

In the EI-mass spectrum of 2 (Table 1) the



\* Experimental conditions for HPLC separation of 1, 2 and 3: Waters Bondapak C 18 column; mobile phase A  $H_2O$  -  $CH_3CN$ , 69: 31 to pH 2 with  $H_3PO_4$ ; mobile phase B  $CH_3CN$ ; elution in a linear gradient

from 10 to 20% B in 10 minutes, followed by a second 10 minutes linear gradient from 20 to 70% B; flow rate 2 ml/minute at room temp; detection 254 nm; retention times (minutes) 5.6 (2), 6.5 (3), 9.7 (1).

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T	m/z (relative intensity)				
Ion	1	2	3	4	
[M] <sup>+</sup> ·	846 (12)	804 (17)	862 (8)	862 (12)	
$[M - CH_3OH]^+$	814 (2)	772 (9)	830 (1)	830 (2)	
$[M - CH_3OH - C_3H_7]^+$	771 (4)	729 (7)	787 (1)	787 (3)	
Ion A	632 (1)	632 (2)	648 (1)	648 (2)	
Ion B	574 (4)	574 (3)	590 (2)	590 (3)	
Ion C	544 (2)	544 (2)	560 (4)	560 (5)	
Chromophore	422 (4)	422 (6)	422 (2)	422 (4)	
$i C_4 H_9 - N_1 = CH_2$	112 (100)	112 (100)	112 (100)	112 (100)	
$\dot{C}H = CH_2$					

Table 1. Main peaks in the EI-mass spectra.\*

\* Data obtained at 70 eV by means of a Varian MAT 311-A spectrometer.

Table 2. <sup>1</sup> H NMR chemical shifts, multiplicity and coupling constants.	Table 2.	<sup>1</sup> H NMR	chemical shifts.	multiplicity and	coupling constants. <sup>a</sup>
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Ductor	δ (ppm) Multiplicity <sup>a</sup>					
Proton -	1	2	3			
NH	8.23 s	8.26 s	8.01 s			
NH-1'	9.03 br s	9.72 br s	8.74 br s			
OH-8	14.40 s	e	14.66 s			
H-13	1.73 s	1.72 s	1.76 s			
H-14	2.31 s	2.28 s	2.30 s			
H-17	5.9~6.4 m	5.8~6.4 m	6.20 $J=2$ , 11 Hz dd			
H-18	5.9~6.4 m	5.8~6.4 m	6.37 $J=11$ , 16 Hz dd			
H-19	5.9~6.4 m	5.8~6.4 m	5.97 $J = 16 \text{ Hz d}$			
H-20	2.2~2.4 m	2.38 m	2.2~2.3 m			
H-21	3.65 m	3.70 $J=9$ Hz d	3.78 s			
OH-21	(3.35) <sup>b</sup> s	(2.95) <sup>b</sup> s	(3.16) <sup>b</sup> s			
H-22	1.7~2.0 m	1.7~2.1 m	1.8~2.0 m			
H-23	2.8~3.0 m	3.30 m	2.8~3.0 m			
OH-23	(3.60) <sup>b</sup> s	(4.15) <sup>b</sup> s	(3.79) <sup>b</sup> s			
H-24	1.42 m	1.7~2.1 m	1.4~1.7 m			
H-25	4.72 J=11, 2 Hz dd	3.54 $J=8$ , 10 Hz dd	4.88 J=10, 2 Hz dd			
OH-25		2.87 $J = 8$ Hz d				
H-26	1.7~2.0 m	1.7~2.1 m	1.4~1.7 m			
H-27	3.25 m	3.35 m	3.34 m			
H-28	5.09 J=8, 13 Hz dd	5.18 $J=10$ , 12 Hz dd	4.99 $J=6$ , 12 Hz dd			
H-29	6.10 J=13 Hz d	5.8~6.4 m	6.06 $J=1$ , 12 Hz dd			
H-30	2.04 s	2.03 s	2.07 s			
H-31	0.82 J=7 Hz d	0.83 $J = 7 \text{ Hz d}$	2.8~3.0 m			
H-32	1.02 $J=7$ Hz d	1.07 $J=7$ Hz d	1.17 $J=7  \text{Hz d}$			
H-33	0.58 $J = 7 \text{ Hz d}$	0.54 $J = 7$ Hz d	0.57 $J=7$ Hz d			
H-34	-0.17 J=7 Hz d	-0.15 J=7 Hz d	-0.13 J=7 Hz d			
H-36	1.99 s		2.04 s			
H-37	3.06 s	3.16 s	3.05 s			
H-4', H-8'	1.6~2.0 m	$1.7 \sim 2.1$ m	1.4~1.7 m			
H-5', H-7'	2.65, 2.90 m	2.64, 2.92 m	1.85, 2.60 m			
H-9′	2.2~2.4 m	2.2~2.3 m	2.2~2.3 m			
H-10′	1.7~2.0 m	1.7~2.1 m	1.8~2.0 m			
H-11', H-12'	0.94 $J = 6.5 \text{ Hz d}$	0.93 $J = 6.5 \text{ Hz d}$	0.92 $J = 6.5$ Hz d			

<sup>a</sup> 200 MHz, CDCl<sub>3</sub> solutions, s=singlet; br s=broad singlet; d=doublet; dd=doublet of doublets; m= multiplet. <sup>b</sup> Values can be interchanged.

° Not detected.

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Carbon		δ (ppm)		Carbon	δ (ppm)		
	1	2	3	atom	1	2	3
1	182.1	181.4	181.5	24	38.1	38.5	37.7
2	114.6	114.9	113.9	25	73.6	71.2	74.0
3	142.7	142.0	142.3	26	38.6	40.0	39.0
4	155.6	155.3	153.3	27	81.2	85.7	79.0
5	112.6	112.0	111.9	28	116.1	113.8	116.5
6	168.4	168.1	168.0	29	145.1	147.8	147.5
7	109.8	110.1	109.2	30	20.5	20.3	20.0
8	171.9	171.1	171.6	31	17.4	17.0	74.7
9	107.9	108.4	107.1	32	11.9	12.5	14.0
10	125.7	125.4	c	33	9.1	8.9	9.5
11	191.8	191.3	191.4	34	11.4	11.4	8.7
12	105.6	105.1	105.0	35	171.9		172.0
13	22.0	22.8	21.3	36	20.5		20.1
14	7.7	7.7	7.4	37	56.5	55.5	56.6
15	168.7	168.6	168.4	2'	95.0	94.9	94.2
16	132.0	132.4	131.1	4'	(36.0) <sup>b</sup>	(35.8) <sup>b</sup>	(35.4) <sup>b</sup>
17	132.7	133.0	133.4	5'	51.7	[51.6] <sup>b</sup>	51.3
18	124.8	123.8	123.2	8'	(36.7) <sup>b</sup>	(36.5) <sup>b</sup>	(36.1) <sup>b</sup>
19	141.2	141.2	143.4	7'	51.7	[51.8] <sup>b</sup>	51.3
20	39.1	39.4	d	9′	66.6	66.5	66.0
21	73.2	72.2	73.7	10'	26.2	26.1	25.6
22	33.7	33.5	33.5	11'	20.9	20.9	20.6
23	77.4	76.7	78.4	12'	20.9	20.9	20.6

Table 3. <sup>13</sup>C NMR chemical shift assignments.<sup>a</sup>

50 MHz,  $C_6D_6$  solutions.

<sup>b</sup> Values in parentheses and brackets can be interchanged.

<sup>c</sup> Overlap by solvent.

<sup>d</sup> Probably overlapped by C-37 signal.

molecular ion is observed at m/z 804; the fragmentation pattern is the same as previously described.<sup>3)</sup> in particular the fragments corresponding to the chromophore and those shown as ions A, B and C (Scheme 1) are unchanged with respect to 1, indicating that the modification is confined to the "ansa" region C-24 to C-29. In the 200 MHz <sup>1</sup>H NMR spectrum of 2 (Table 2) the signal of CH<sub>3</sub>-36 ( $\delta$ : 1.99 ppm in 1) is missing and H-25 shifts upfield from 4.72 (value observed for 1) to 3.54 ppm. The coupling (J=8 Hz) with the OH signal at 2.87 ppm is clearly observable. The <sup>1</sup>H chemical shift values for 1, also reported in Table 2, are completely consistent with the structure and in agreement with the reported values for rifamycin S.<sup>4)</sup> The signals of C-35 and C-36 (171.9 and 20.5 ppm respectively, in 1) are missing from the <sup>13</sup>C NMR spectrum of 2 (Table 3) and the C-25 peak (73.6 ppm in 1) is observed at 71.2 ppm, whereas all other signals are found as expected and in agreement with literature values<sup>5)</sup> and

known substituent effects. The data allow one to assign to **2** the structure of the 25-*O*-deacetyl derivative of LM 427; the same deacylation process was reported for rifampicin.<sup>6)</sup> The assignment has been confirmed by direct comparison with an authentic sample prepared<sup>1)</sup> from 25-*O*-deacetylrifamycin S.<sup>7)</sup>

As far as compound 3 is concerned, its EImass spectrum shows the molecular ion at m/z862, *i.e.*, the molecular weight is 16 mass units higher than that of the parent compound. The same increase of 16 mass units is observed for ions A, B and C, while the chromophore ion is unchanged (Table 1). The above data indicate that an oxygen atom has been introduced in the "ansa" segment C-15 to C-20. In the <sup>1</sup>H NMR spectrum (Table 2) the CH<sub>3</sub>-31 doublet (0.82 ppm for 1) is missing whereas an additional signal corresponding to a CH<sub>2</sub>OH group is found in the range 2.8~3.0 ppm, badly overlapped by the H-23 multiplet. The presence of the additional CH<sub>2</sub>OH group causes noticeable conformational changes in that region of ansa chain. Coupling constants different from the reported values<sup>4)</sup> are observed for H-19 (d,  $J_{19,18}$ =16 Hz) and H-21, which appears (both in CDCl3 and in  $C_{\theta}D_{\theta}$ ) as a broad singlet at 3.78 ppm. For this reason it has not been possible to detect a sizeable correlation peak between CH<sub>2</sub>OH-31 and H-21 in a 2D homo-correlation experiment carried out on a  $C_6D_6$  solution of 3. Similar conformational effects were not observed for 3,31-dihydroxyrifamycin S in CDCl<sub>3</sub> solution.<sup>8)</sup> In the <sup>13</sup>C NMR spectrum, the C-31 signal (17.4 ppm in 1) is missing, and an additional peak appears in a region appropriate for a CH<sub>2</sub>OH group (74.7 ppm). On the basis of the above data, the structure of the 31-hydroxyl derivative of LM 427 is assigned to compound 3.

A third, minor metabolite (4) was also isolated; its EI-mass spectrum shows the molecular ion at m/z 862, indicating that also in this case an oxygen atom has been introduced in the molecule. The fragmentation pattern shows that the chromophore is unchanged; therefore the additional oxygen again must be located in the "ansa" portion. Due to the minimal amount of 4 it has not been possible to perform a complete NMR analysis, however, the <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>) showed the absence of the olefinic C-30 methyl singlet and the presence of all the other "ansa" methyl signals. Therefore, we suggest for 4 the structure of the 30hydroxyl derivative of 1. These results indicate that the hydroxylation pattern in humans parallels to some extent the microbial oxidation at the C-30 and C-31 methyl groups.8,10)

The biological activity of **2** is quite similar to that of LM 427, whereas metabolites **3** and **4** show less activity.<sup> $\theta$ </sup>

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